

A HELIANGOLIDE FROM *SCHKUHRIA PINNATA**

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Key Word Index—*Schkuhria pinnata*; Compositae; sesquiterpene lactones; heliangolides.

Abstract—Re-investigation of the aerial parts of *Schkuhria pinnata* afforded in addition to the heliangolides isolated previously a new one, the structure of which followed from the spectroscopic data.

From the aerial parts of *Schkuhria pinnata* (Lam.) O. Ktze (Compositae, tribe Heliantheae) so far, in addition to widely distributed compounds, the two heliangolides 1 and 3 were reported [1], 1 being present also in *S. virgata* [2]. Furthermore, the unusual aromatic compound schkuhrianol has been isolated [3]. A re-investigation afforded in addition to 1, 3 and 7 [4, 5] a further heliangolide, which, however, could only be isolated as its diacetate 6. The molecular formula $C_{29}H_{36}O_{11}$ and the 1H NMR data (Table 1) indicated that a dehydro compound of the diacetate of 3 was present. While most signals were nearly identical with those of 2 and 4, the ester residue at C-3 was obviously different. A quartet of quartets at 3.26 ppm (in C_6D_6), had to be assigned to the proton of an isopropyl group, which was coupled with methyl groups (1.17 d and 1.06 d). The chemical shift and the mass spectrum, where loss of $C_5H_8O_3$ could be recognized, indicated the presence of a 2-oxo-isovalerate, while the second ester residue obviously was the same as

in 2 and 4. Though the distribution of the two ester groups could not be established, biogenetic considerations suggest that the same group is attached to C-8 in all three heliangolides.

EXPERIMENTAL

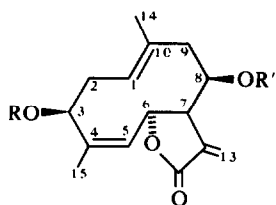
The fresh aerial parts (Botanic Gardens Berlin-Dahlem, voucher 80/1393 A) was extracted with Et_2O -petrol, 1:2 and the resulting extract was separated first by CC (Si gel). The polar fractions (Et_2O -MeOH, 20:1) were further separated by TLC (Si gel, $CHCl_3$ - C_6H_6 - Et_2O , 1:1:1) affording 30 mg 1, 10 mg 3, 2 mg 7 and a mixture of 1, 3 and 5, which after acetylation could be separated by HPLC (reversed phase, MeOH- H_2O , 3:2). Finally, in addition to 2 and 4, 2.5 mg 6 were obtained, colourless gum, IR $\nu_{max}^{CCl_4}$ cm^{-1} : 1770 (γ -lactone), 1745, 1225 (OAc), 1730 ($C=CCO_2R$, $-COCO_2R$); MS m/z (rel. int.): 560.226 $[M]^+$ (0.5) ($C_{29}H_{36}O_{11}$), 444 $[M - RCO_2H]^+$ (0.5), 344 $[M - R'CO_2H]^+$ (1), 329 $[344 - 'Me]^+$ (3), 229 $[344 - RCO_2]^+$ (44), 228 $[344 - RCO_2H]^+$ (44), 157

Table 1. 1H NMR spectral data of compounds 2 and 6 (400 MHz, C_6D_6)

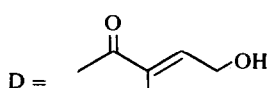
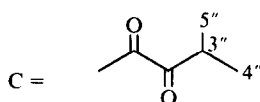
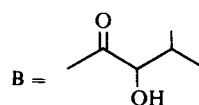
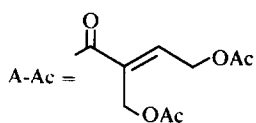
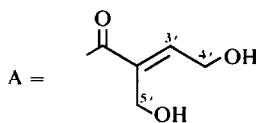
| | 2 (77°) | 6 (50°) | | 2 | 6 |
|------------------|----------------------|------------|--------------------|--------|---------|
| H-1 | 4.91 m | 4.83 m | H-3' | 7.05 t | 7.05 t |
| H-2 | 2.50 ddd | 2.44 br dd | H-M ₁ ' | 4.72 d | 4.95 dd |
| H-2' | 1.94 ddd | 1.91 m | H-4 ₂ ' | | 4.85 dd |
| H-3 | 5.18 dd | 5.04 br dd | H-5 ₁ ' | 4.83 s | 5.11 d |
| H-5 | 4.91 ddq | 4.81 br d | H-5 ₂ ' | | 5.06 d |
| H-6 | 5.89 dd | 5.86 br d | H-3'' | — | 3.26 qq |
| H-7 | 2.45 br s | 2.28 br s | H-4'' | — | 1.17 d |
| H-8 | 5.24 ddd | 5.21 dd | H-5'' | — | 1.06 d |
| H-9 | 2.10 dd | 2.0 br d | OAc | 1.82 s | 1.83 s |
| H-9' | 2.69 dd | 2.76 br dd | | 1.72 s | 1.74 s |
| H-13 | 6.27 d | 6.25 d | | 2.07 s | |
| H-13' | 5.28 d | 5.18 d | | | |
| H-14 } H-15 } | 1.58 d } 1.50 d } | | | | |

J (Hz): compound 6: 1,2 = 9.5; 2,2' = 14; 5,6 = 10; 7,13 = 2.5; 7,13' = 2; 8,9 = 8.9'
= 3; 9,9' = 14; 3',4' = 6; 4₁,4₂ = 15; 5₁,5₂ = 12.5; 3'',4'' = 3'',4'' = 7.

*Part 373 in the series "Naturally Occurring Terpene Derivatives". For Part 372 see Bohlmann, F. and Gupta, R. K. (1981) *Phytochemistry* 20, (in press).



| | | | | | | | |
|----|----|------|---|------|---|------|----|
| | 1 | 2 | 3 | 4 | 5 | 6 | 7 |
| R | Ac | Ac | B | B | C | C | Ac |
| R' | A | A-Ac | A | A-Ac | A | A-Ac | D |



$[\text{HOCH}_2\text{CH}=\text{C}(\text{OAc})\text{CO}]^+$ (100), 115 $[\text{157} - \text{ketene}]^+$ (95),
71 $[\text{C}_3\text{H}_7\text{CO}]^+$ (71).

$$[\alpha]_{24}^1 = \frac{589 \quad 578 \quad 546 \quad 436 \quad 365 \text{ nm}}{-62 \quad -63 \quad -73 \quad -134 \quad -218}$$

($c = 0.21$, CHCl_3).

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